

Effects of Immersion Freezing on Simulations of Mixed-Phase Stratus Clouds (Theory and Results)

Gijs de Boer

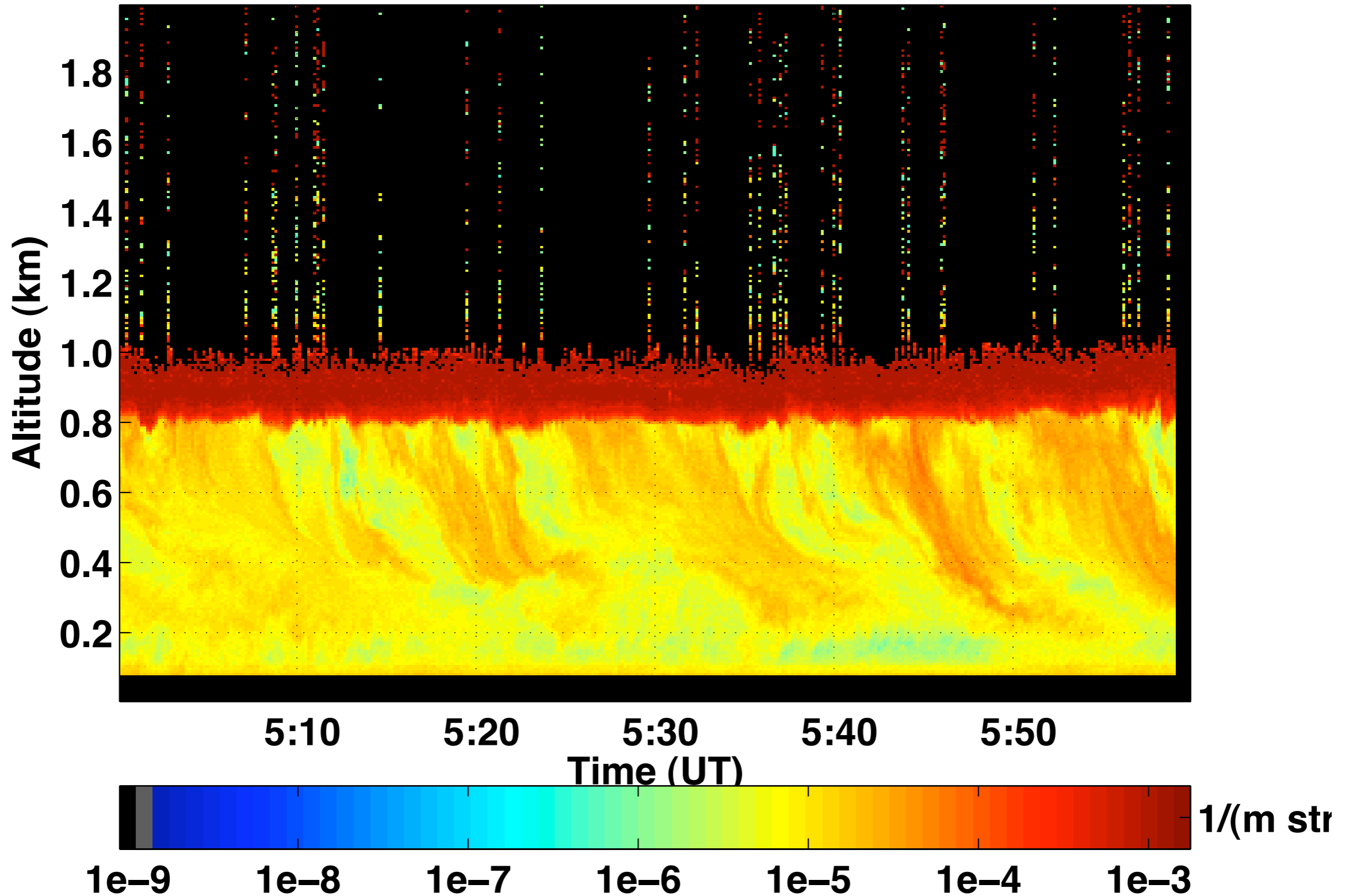
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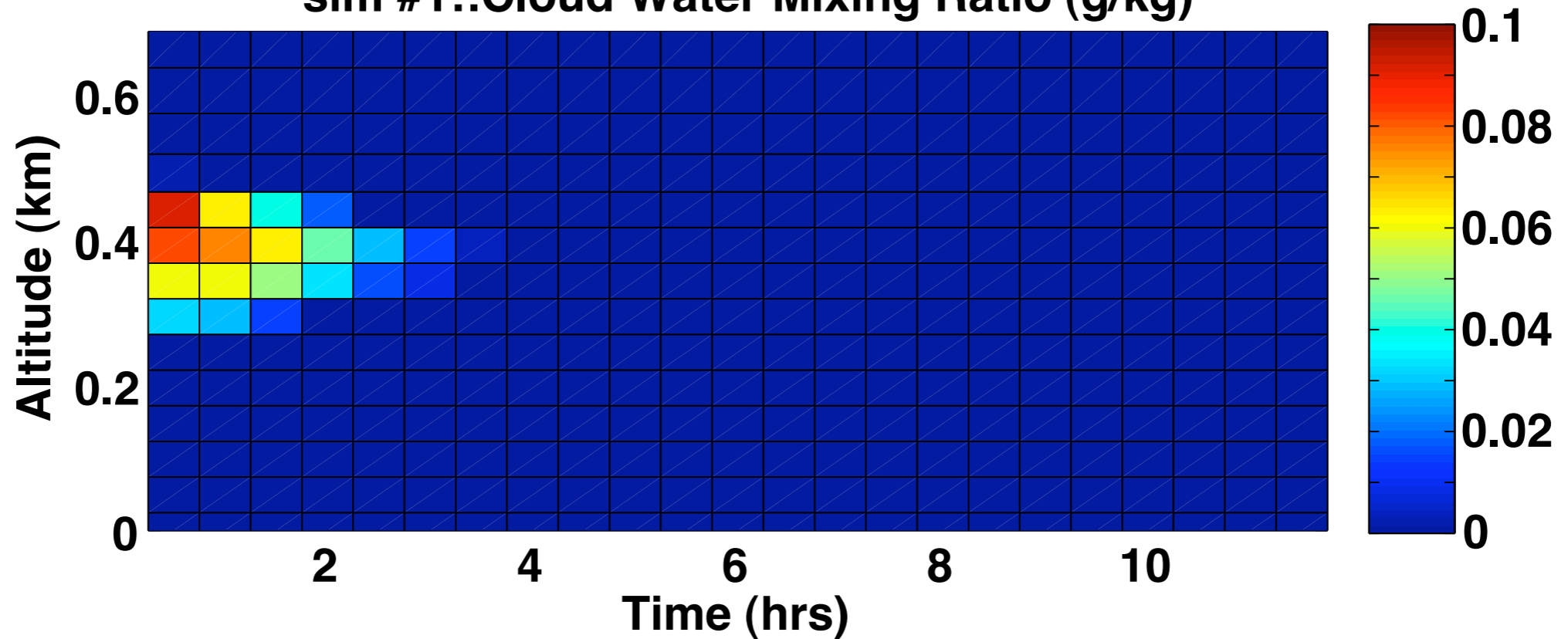
Introduction

Lidar backscatter cross section (Masked values shown in black and white)

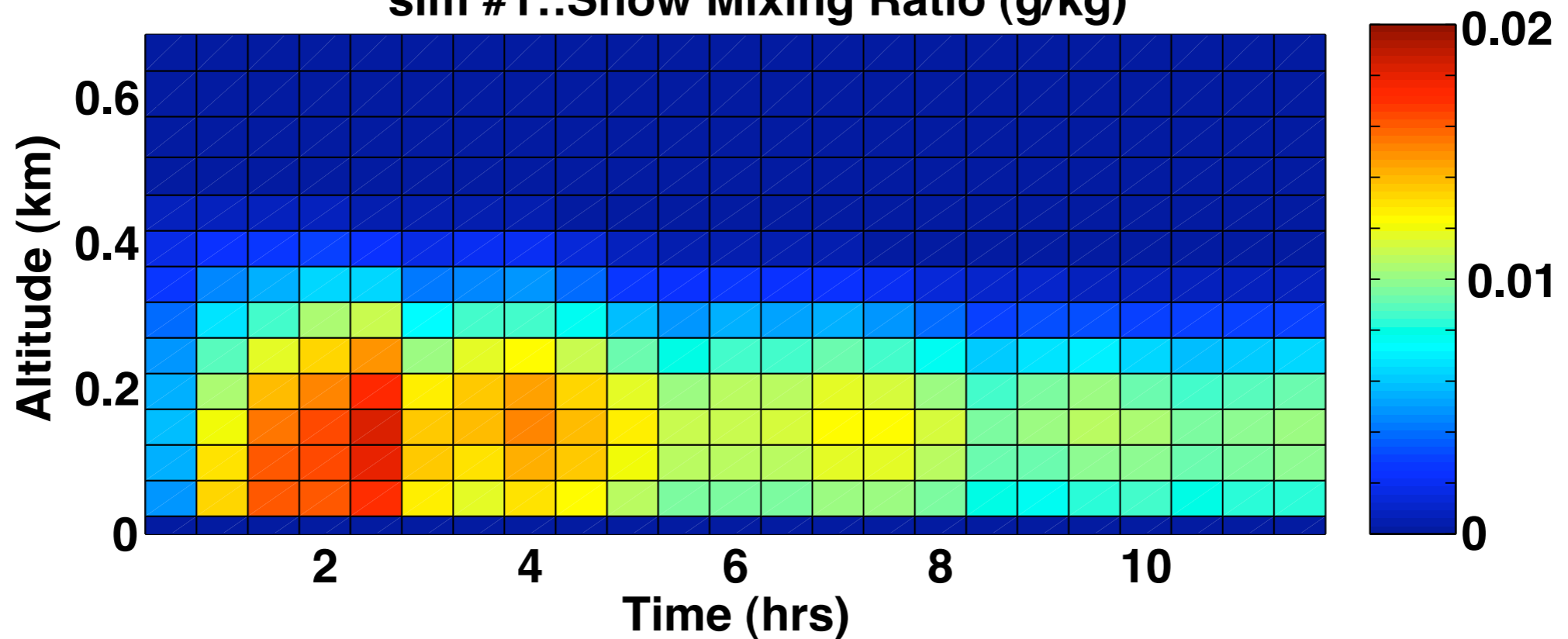


Introduction

sim #1::Cloud Water Mixing Ratio (g/kg)

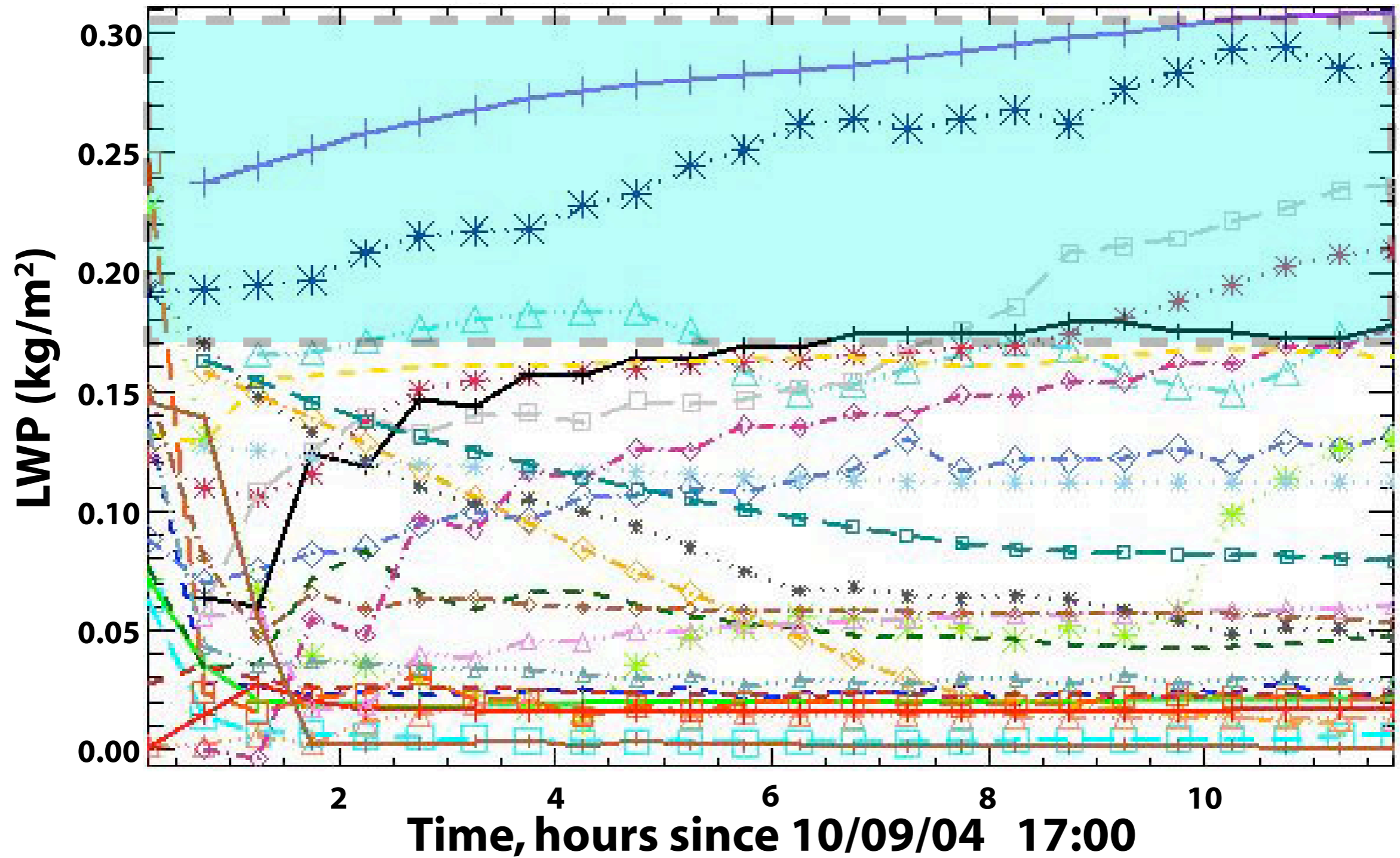


sim #1::Snow Mixing Ratio (g/kg)

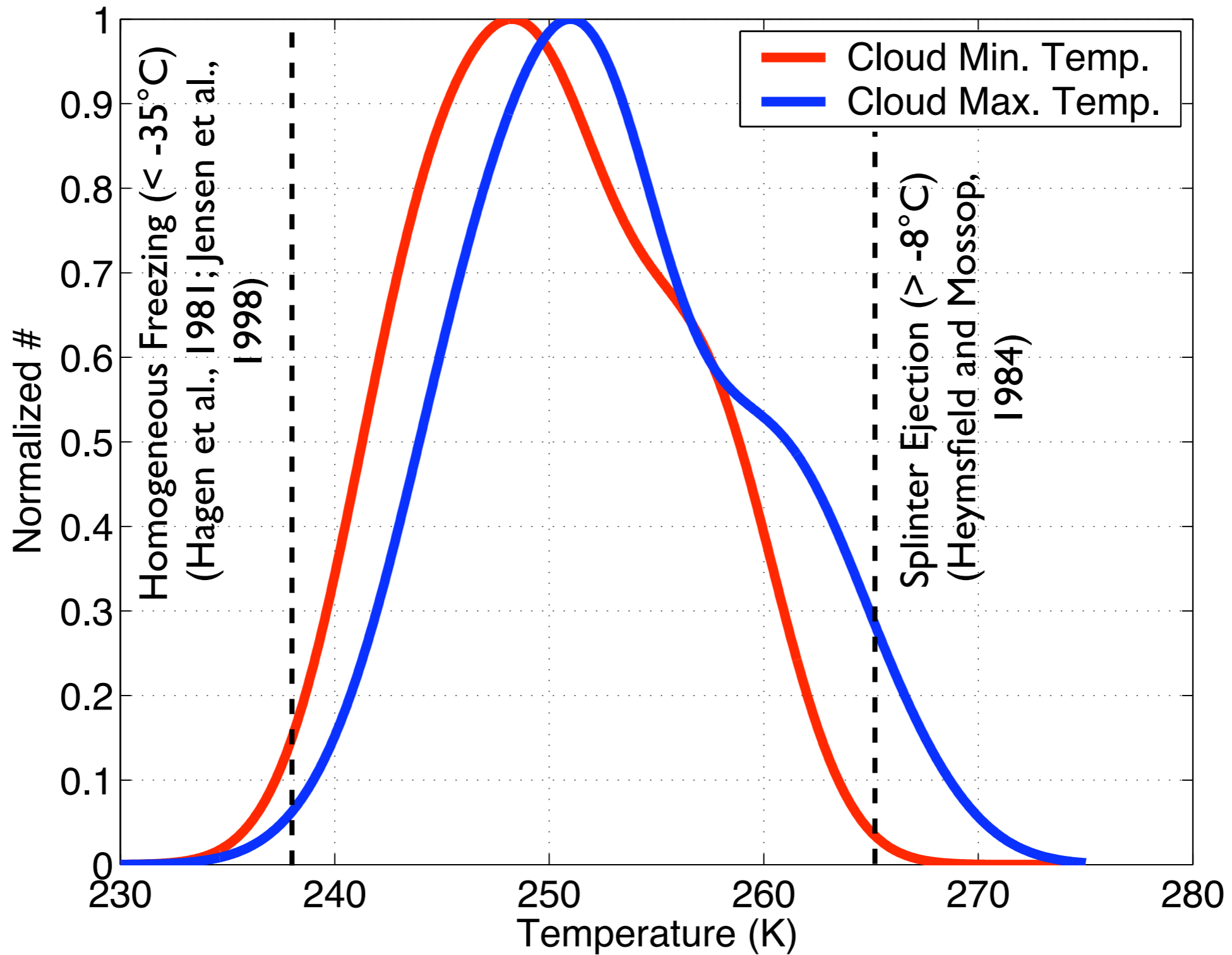


Introduction

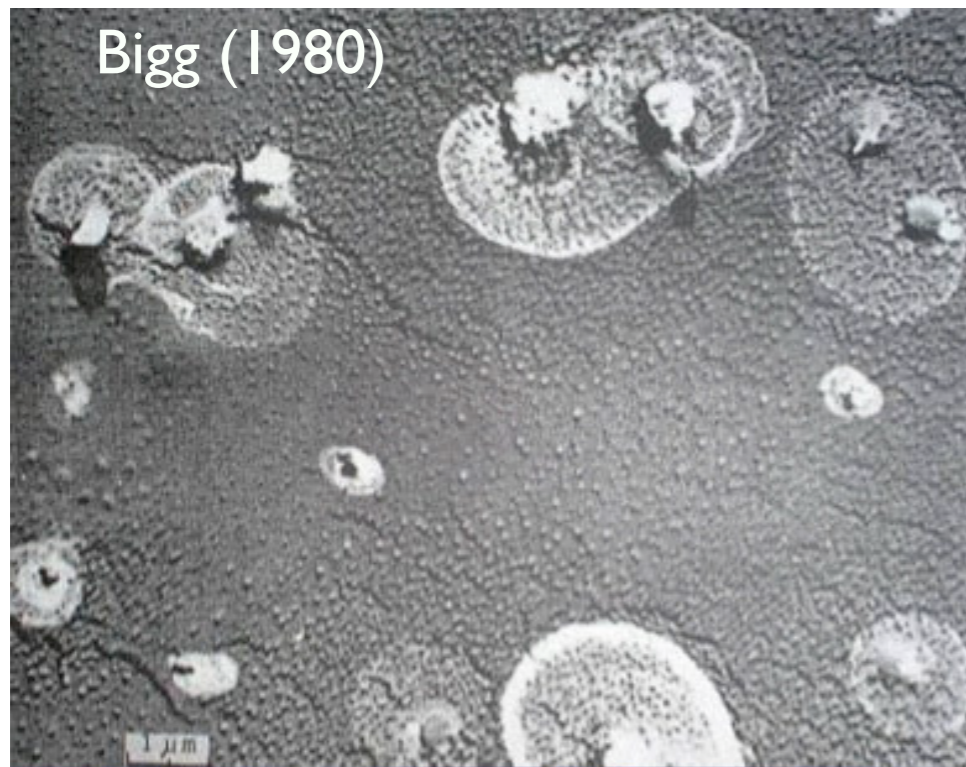
Cloud liquid water path, Case: b1



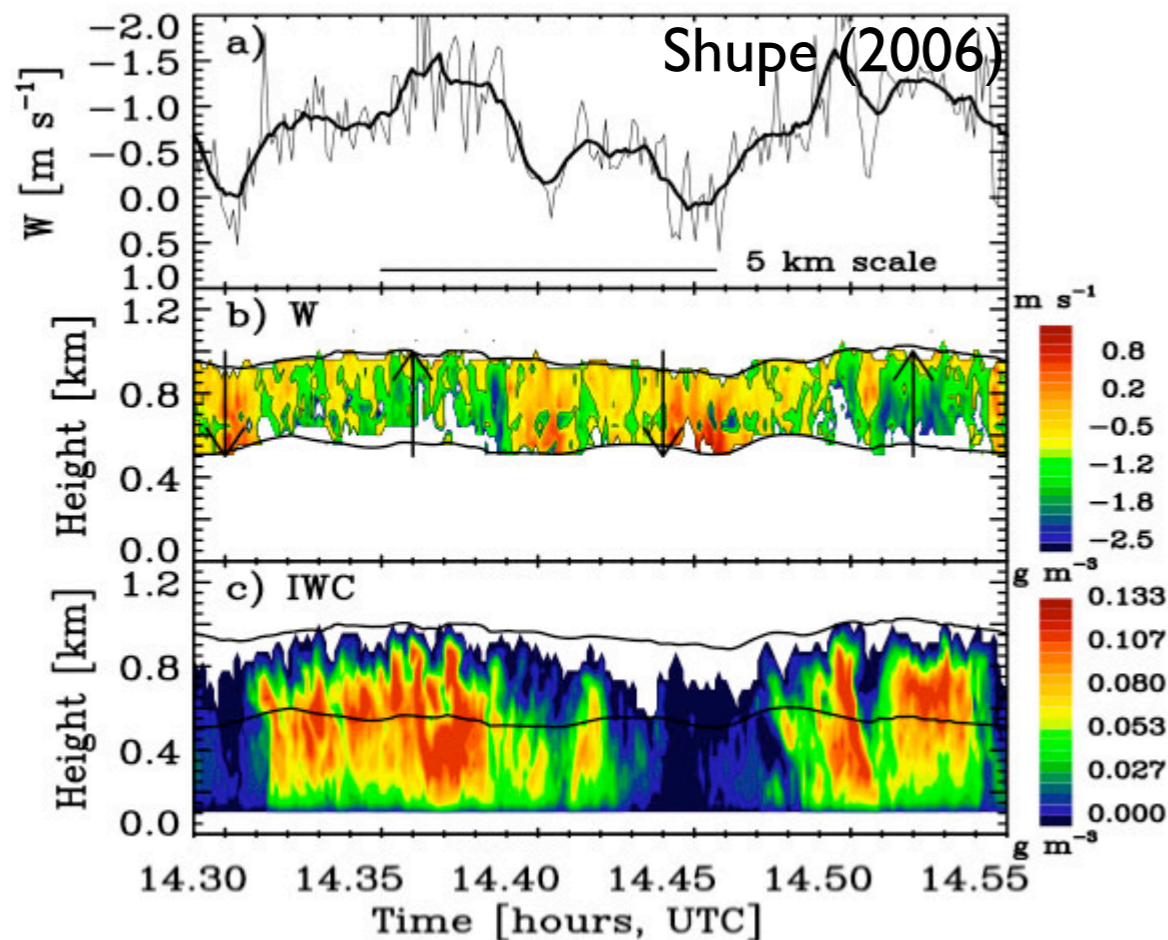
Introduction



Introduction



- Bigg (1980) observed sulfuric acid coating on aerosol particles during winter
- Sulfuric coating is water soluble, transforming possible IN into CCN.
- Blanchet (2007) hypothesizes that sulfur coating is a result of sulfur emissions from Siberia, and that resulting particles in Arctic have reduced ice nucleating ability.



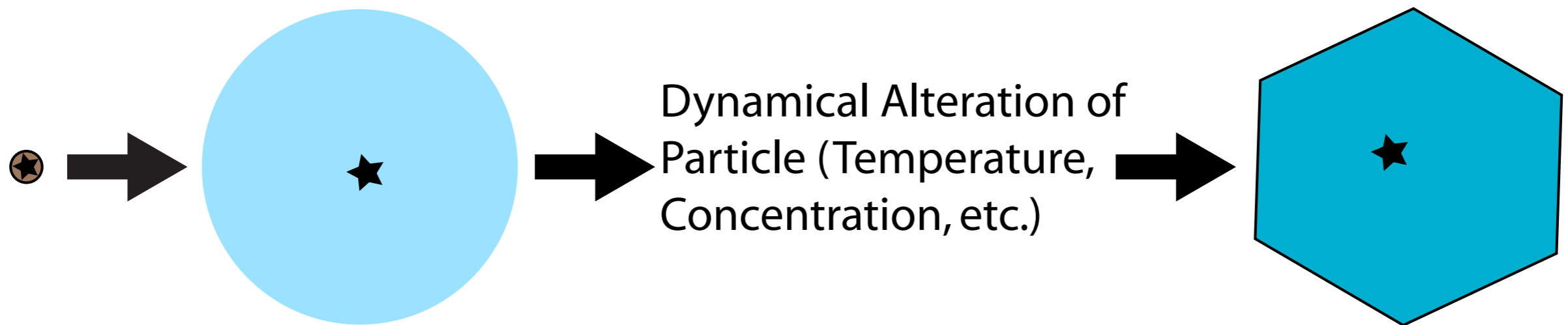
From ground-based sensors:

- Large increases in IWC in updrafts
- Decrease in Liquid Fraction in updrafts

From in-situ measurements:

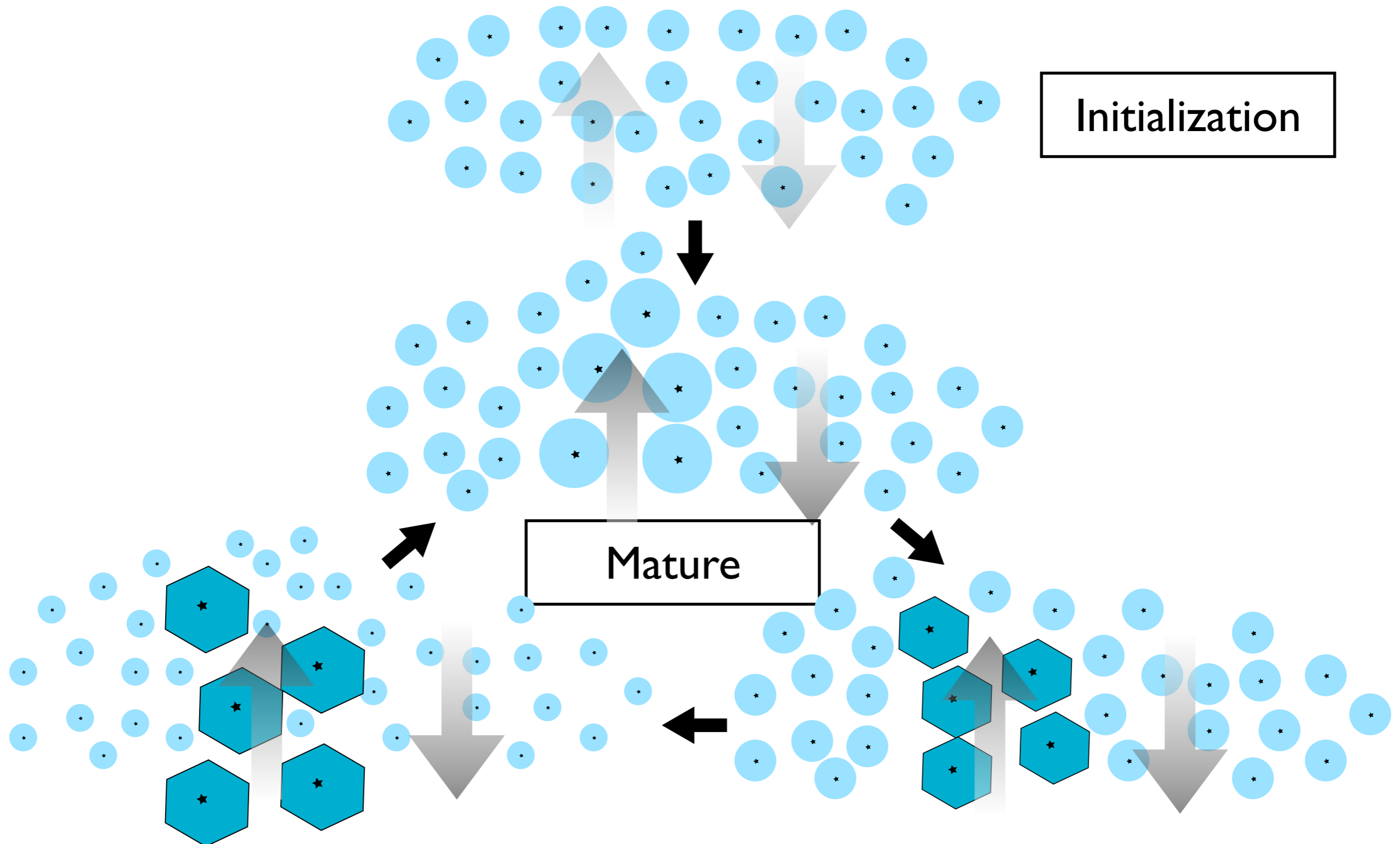
- Ice crystal concentrations strongly proportional to concentration of drops larger than $20 \mu\text{m}$. (Rangno & Hobbs, 2001)

Immersion Freezing

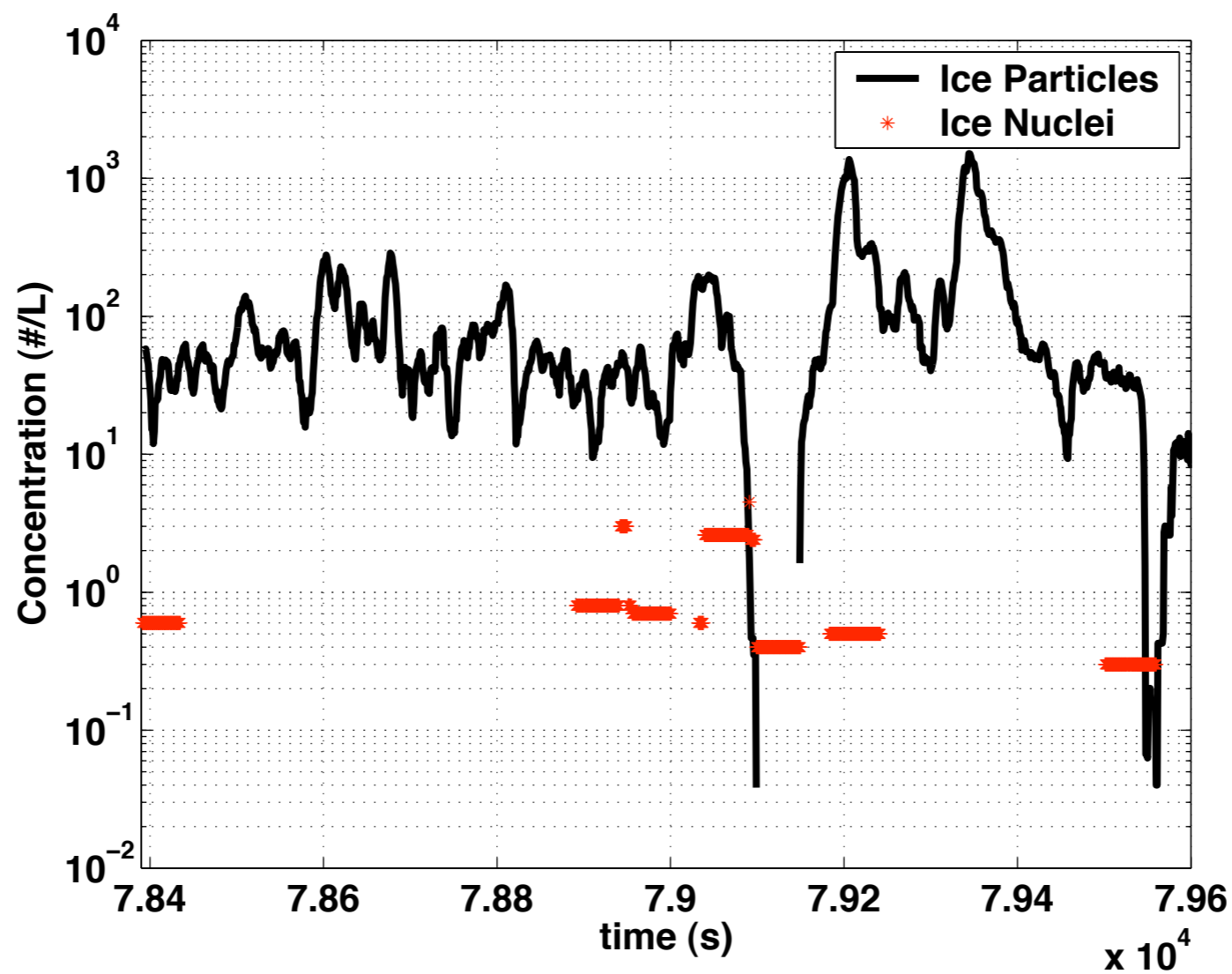
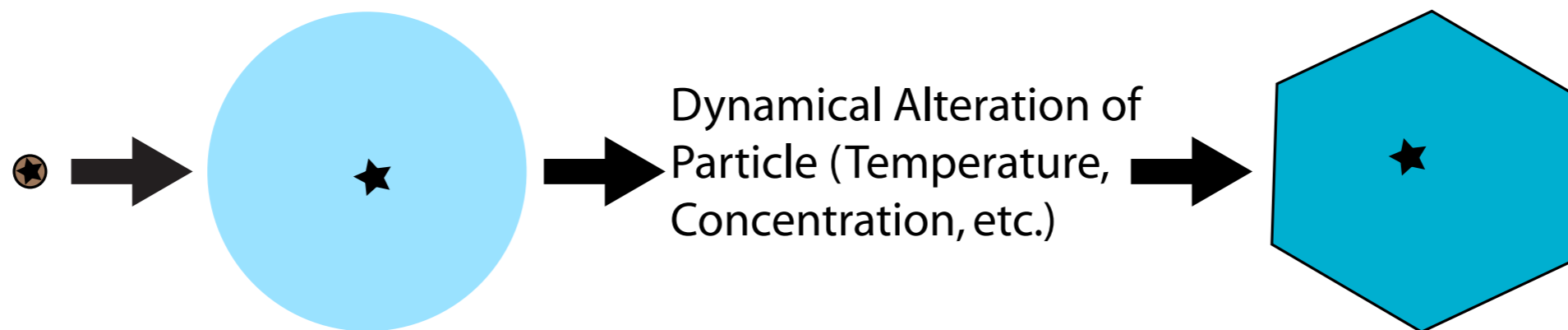


Immersion Freezing

Conceptual Model for Mixed-Phase Stratus

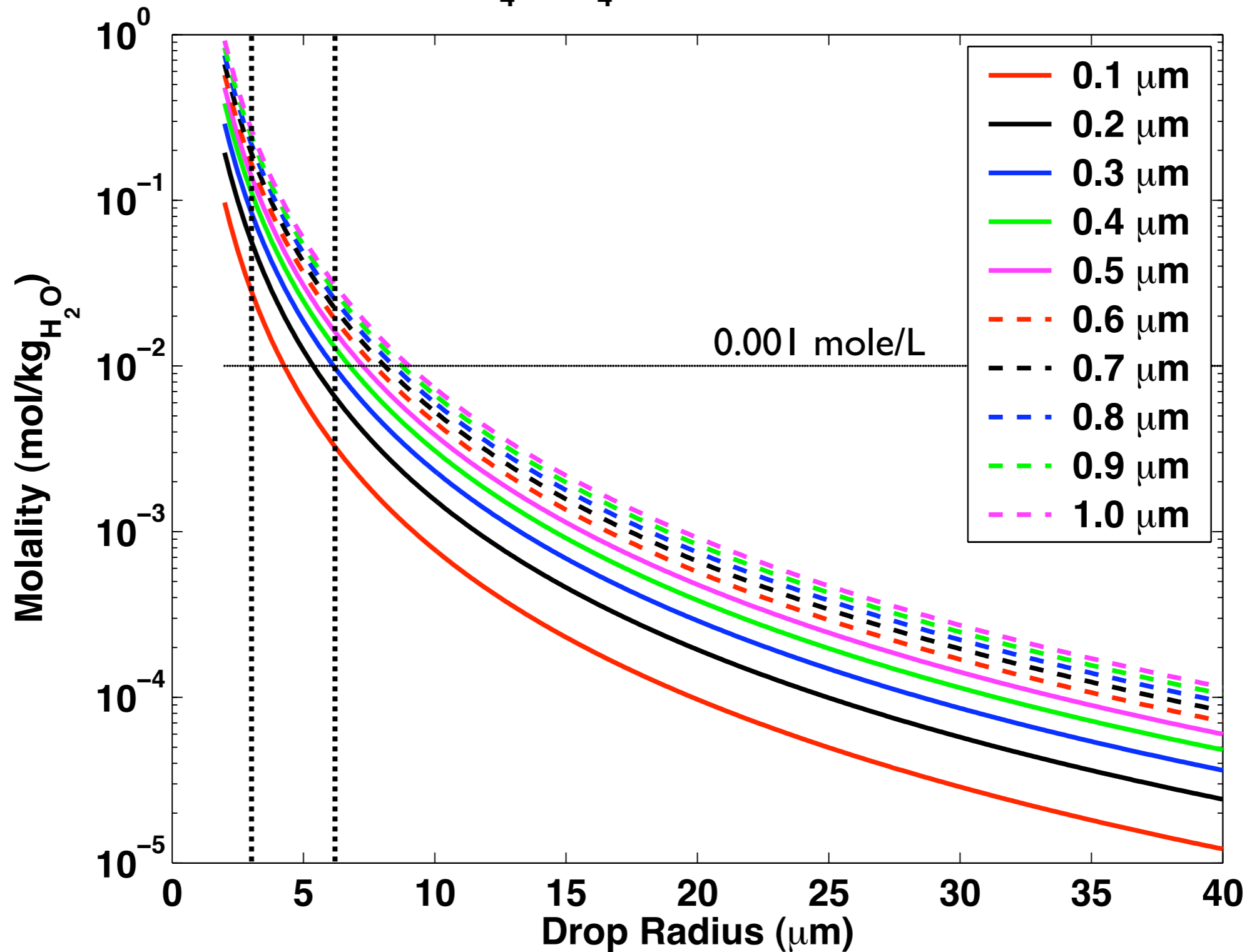


Immersion Freezing



Immersion Freezing

NH_4HSO_4 , Illite, 70% Soluble



Simulations

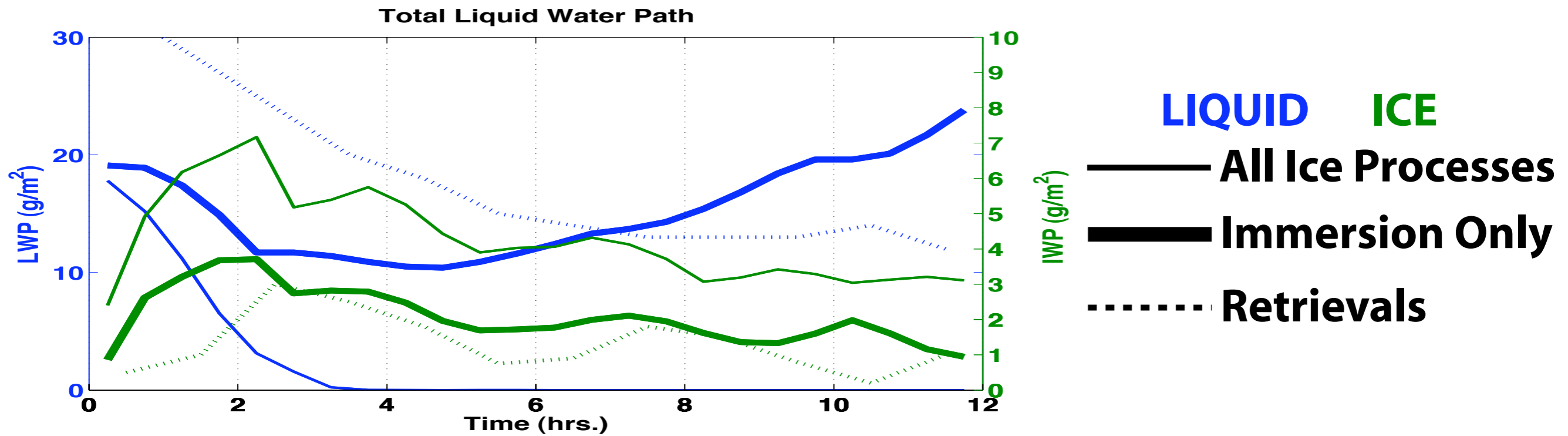
2-D Simulations for the SHEBA mixed-phase case

- University of Wisconsin Non-Hydrostatic Modeling System (UW-NMS: Tripoli, 1992)

- Advanced Microphysical Prediction System (AMPS: Hashino and Tripoli, 2007)
 - Size-resolved liquid and ice microphysics with diagnostic aerosol (IN and mixed).
 - Immersion freezing: Reisin (1996), and solubility effect from Diehl & Wurzler (2004)
 - CCN: 70% soluble NH_4HSO_4 mixed particle

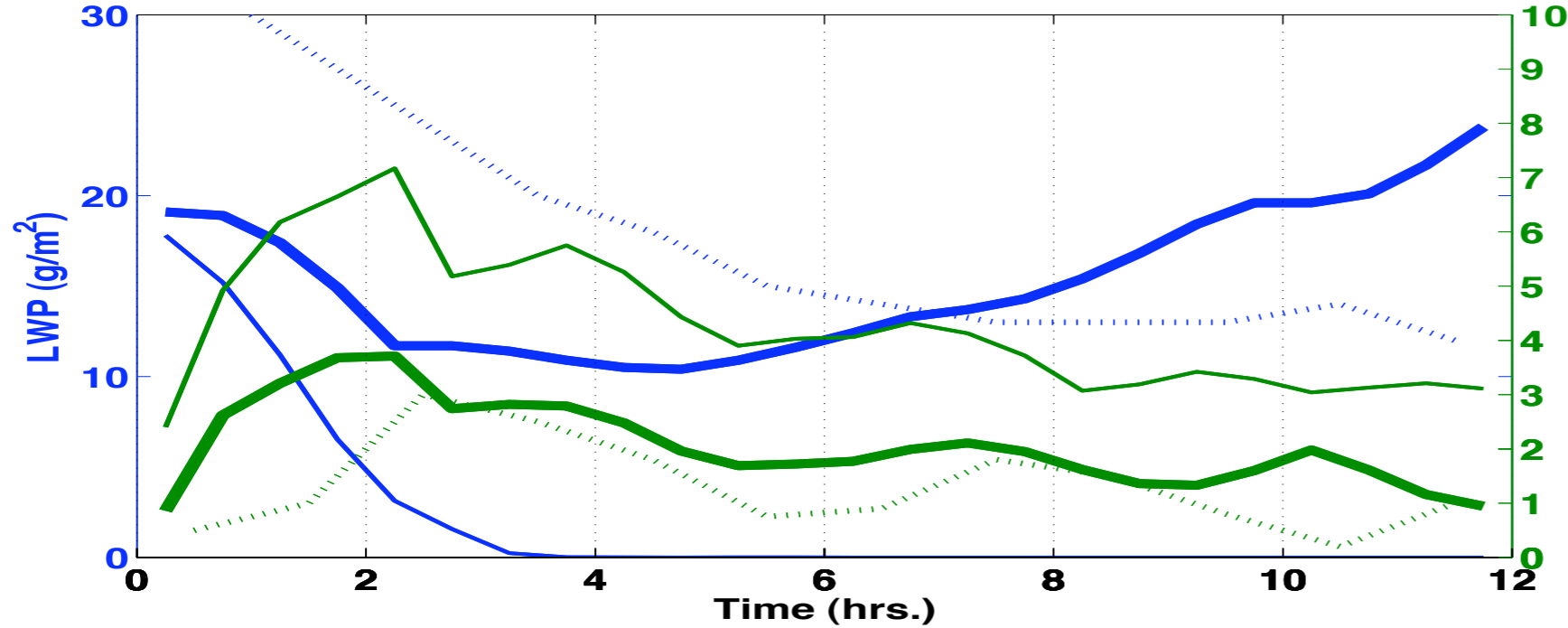
- Resolution: 200 m horizontal, 50 m + vertical

Simulations



Simulations

Total Liquid Water Path

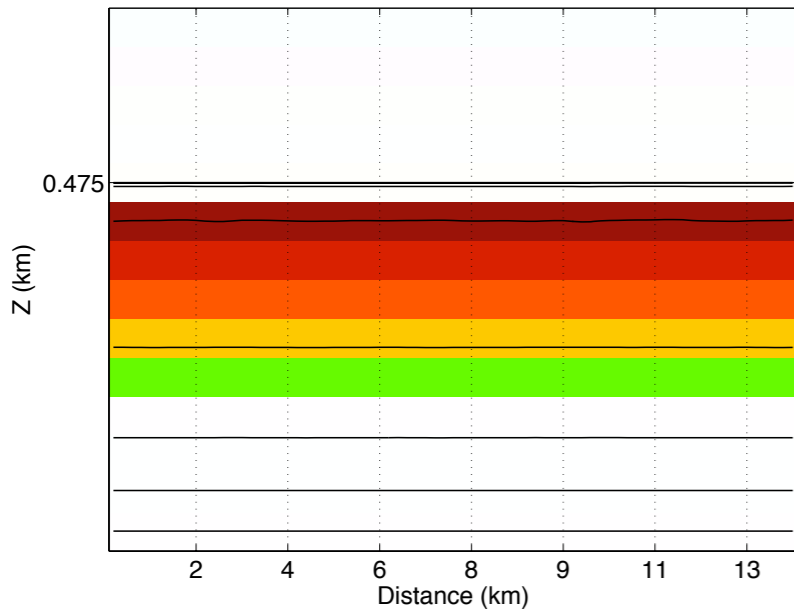


10 minutes

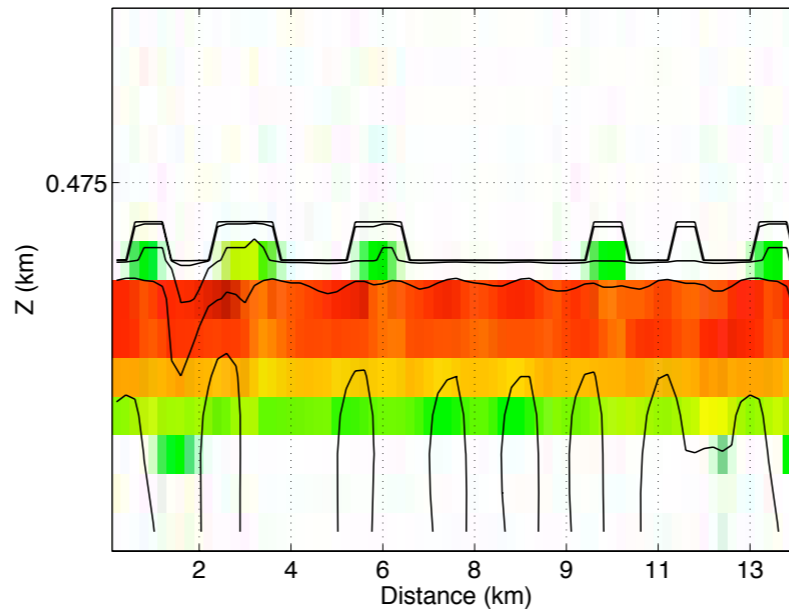
360 minutes

720 minutes

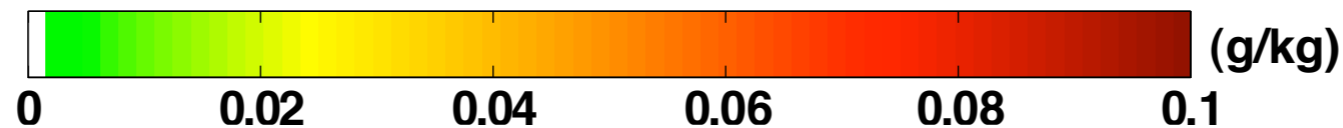
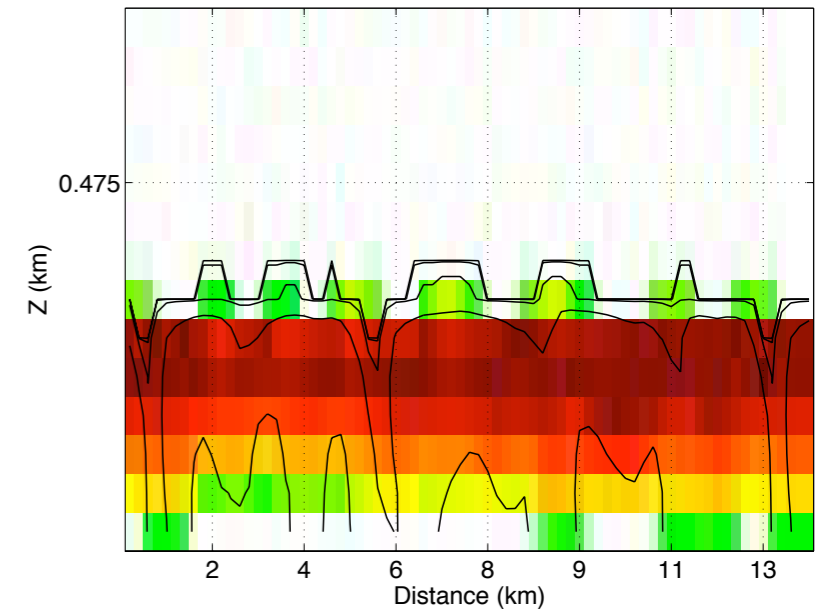
QC (Shaded), QI(contours): 07-May-1998 09:10:00



QC (Shaded), QI(contours): 07-May-1998 15:00:00

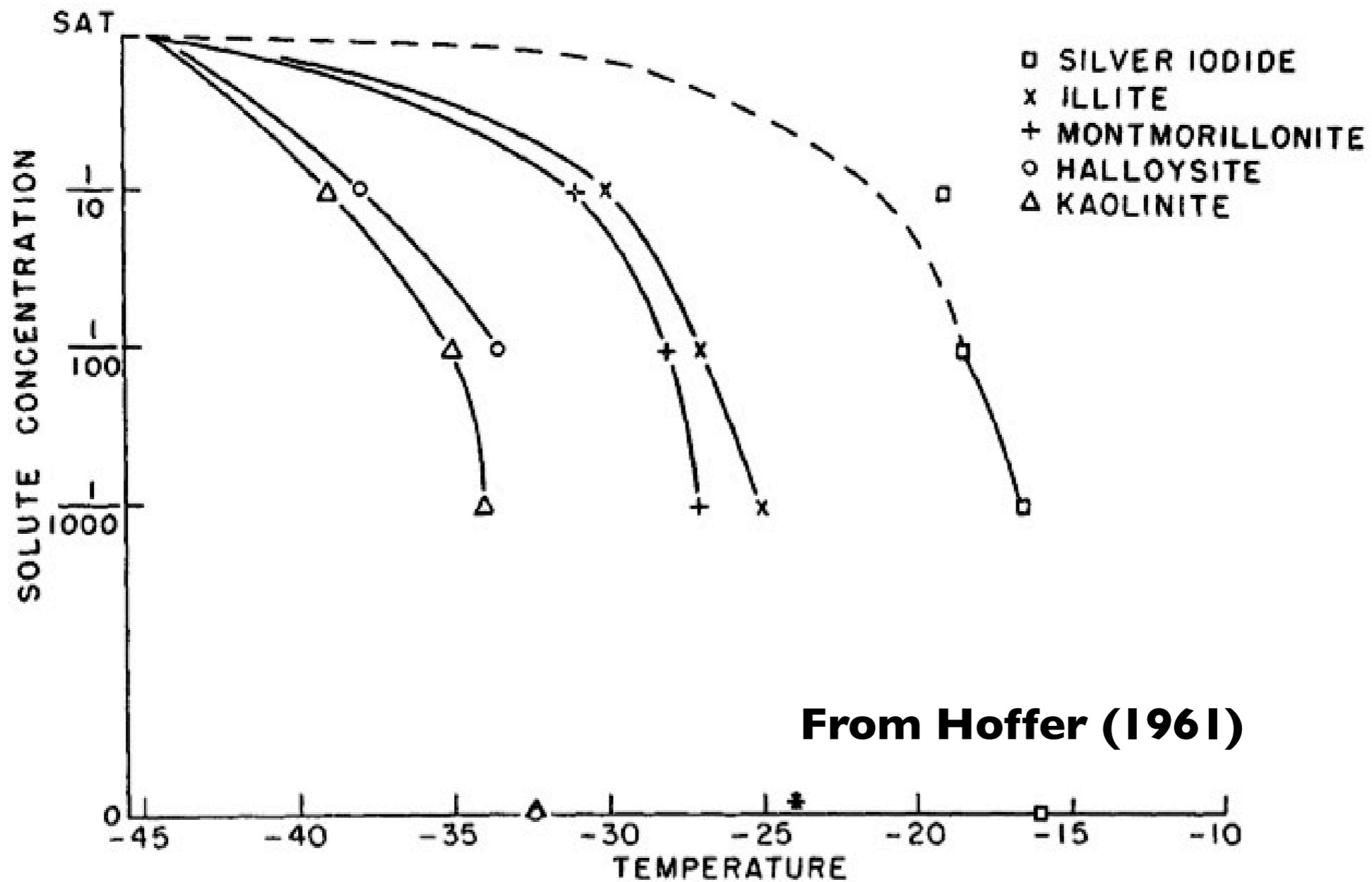


QC (Shaded), QI(contours): 07-May-1998 21:00:00



Simulations

Effect of Solution Concentration



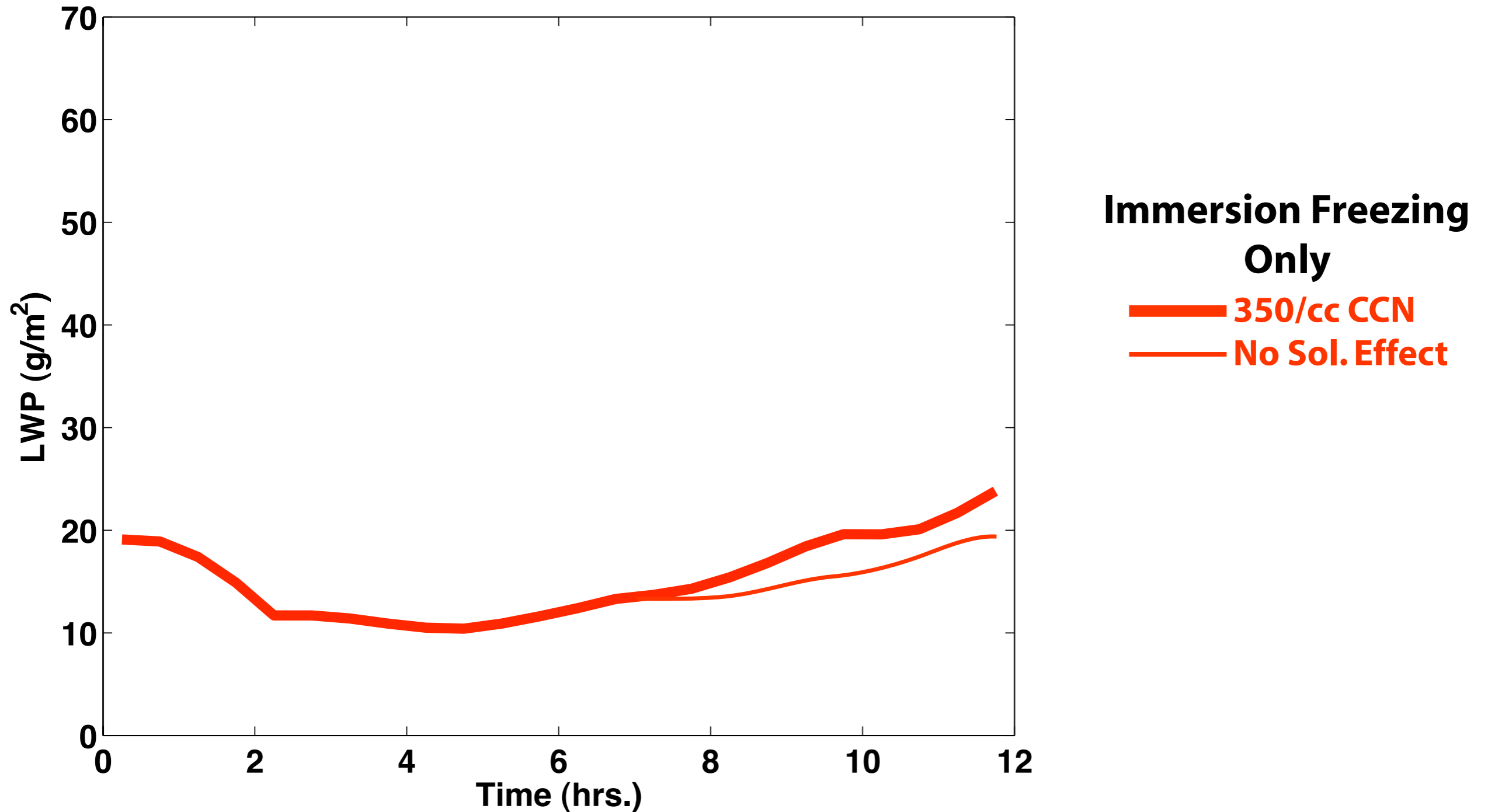
From Hoffer (1961)

FIG. 5. Median droplet freezing temperature for droplets of different solute concentration (100-120 microns diameter).

Simulations

Effect of Solution Concentration

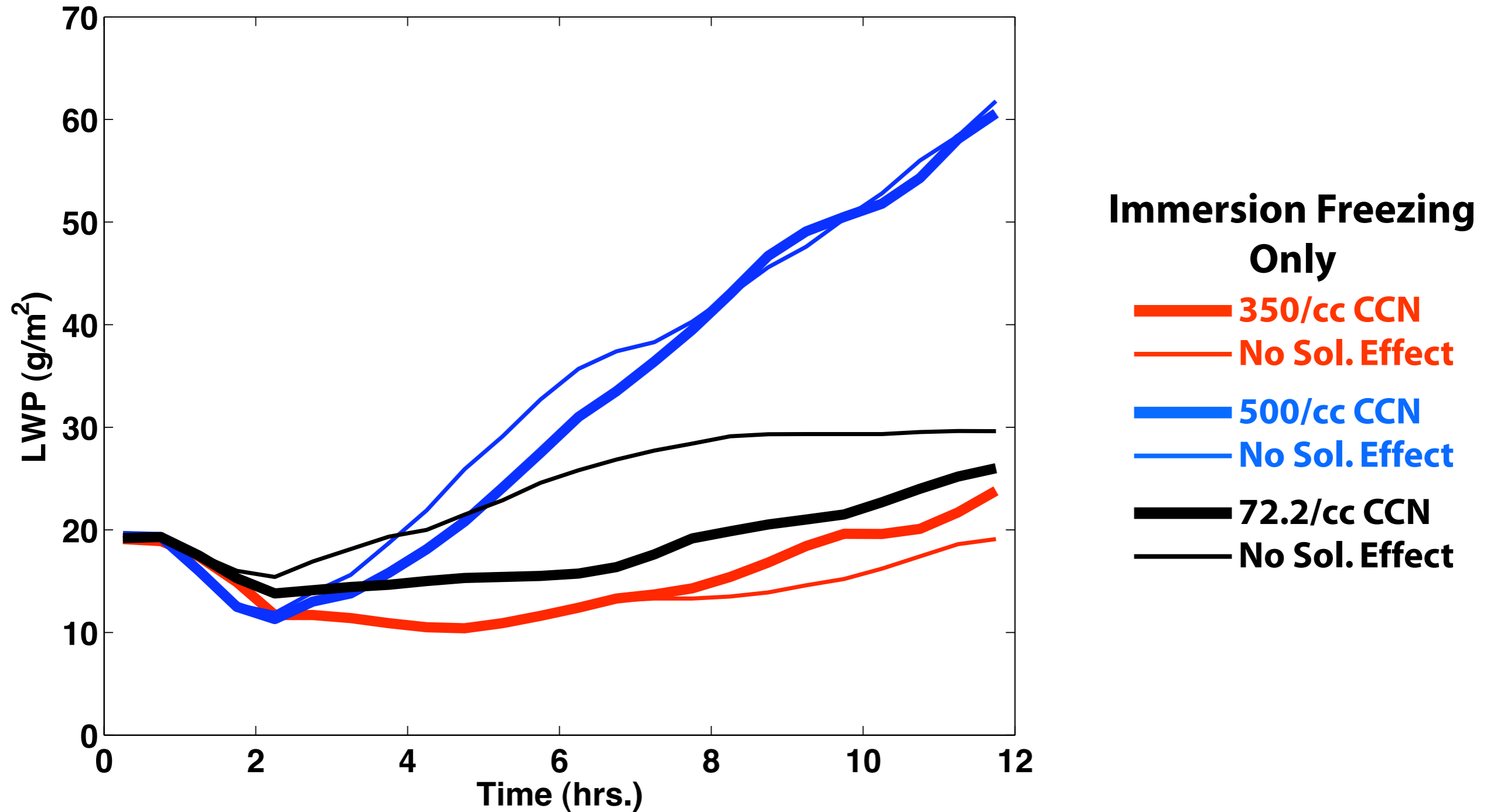
Total Water Path



Simulations

Effect of Solution Concentration

Total Water Path



Summary

- Understanding of ice nucleation mechanisms is key for understanding and modeling mixed-phase stratus lifetime.
- Observation of IN involved with immersion freezing is uncertain at this time.
- Simulations with only immersion freezing active produce approximately the correct amount of ice in high resolution simulation.
- The freezing point depression due to presence of soluble material inside droplets significantly affects simulations, but pathways for these effects are not necessarily straightforward.